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Hirabai Haridas Vidyanagari, Amrutdham, Panchavati, Nashik - 422 003. (Maharashtra) India.

雷: 0253 - 2221121, 2517003, 2510262 Web: www.pharmacy.kkwagh.edu.in Email: principal-bpharmacy@kkwagh.edu.in, disp-bpharmacy@kkwagh.edu.in

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3.3.2

Number of books and chapters in edited volumes/books published and papers published in national/ international conference proceedings per teacher during last five years



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1. Cover page, content page and first page of the selected publication.



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#### 3.3. Research Publications and Awards

# 3.3.2 Number of books and chapters in edited volumes/books published and papers published in national/international conference proceedings per teacher during last five year

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1	<b>Dr. Vaibhav G. Bhamre</b> , Mr. Shivraj P. Jadhav	Concepts of Physical Pharmaceutics I	National	2021-2022	97893928673 92	K. K. Wagh College of Pharmacy	PharmaCareer Publications
2	Bhamare V. G., Amrutkar R. D., Patil V, Upasani C. D.	Growing impact of herbal bioenhancers in pharmaceutical industries	International	2021-2022	97831107467 92	K. K. Wagh College of Pharmacy	De-Gruyter
3	Shamal. D. Dawange	A Laboratory Manual for Pharmacognosy	National	2021-2022	978819540742 2	K. K. Wagh College of Pharmacy	Briliant Publication
4	Dr. K. S. Jain, D. K. Kadam	A Practical of Pharmaceutical Analysis-	National	2020-2021	97893905968 98	K. K. Wagh College of Pharmacy	Nirali Prakashan
5	Dr. K. S. Jain, D. K. Kadam, K. P. Baviskar	Pharmaceutical Analysis I	National	2019-2020	978-93- 90506-60-6	K. K. Wagh College of Pharmacy	Nirali Prakashan
6	Dr. K. S. Jain	A Practical book of Pharmaceutical Organic Chemistry	National	2018-2019	978-93- 87686-05-2	K. K. Wagh College of Pharmacy	Nirali Prakashan
7.	Dr. K. S. Jain	Pharmaceutical Organic Chemistry– I, Simplified	National	2018-2019	978-93- 87397-62-0	K. K. Wagh College of Pharmacy	Nirali Prakashan
8	Dr. K. S. Jain, M.G.Shinde	Practical Pharmaceutical Inorganic Chemistry simplified	National	2018-2019	978-93- 88293-31-0	K. K. Wagh College of Pharmacy	Nirali Prakashan
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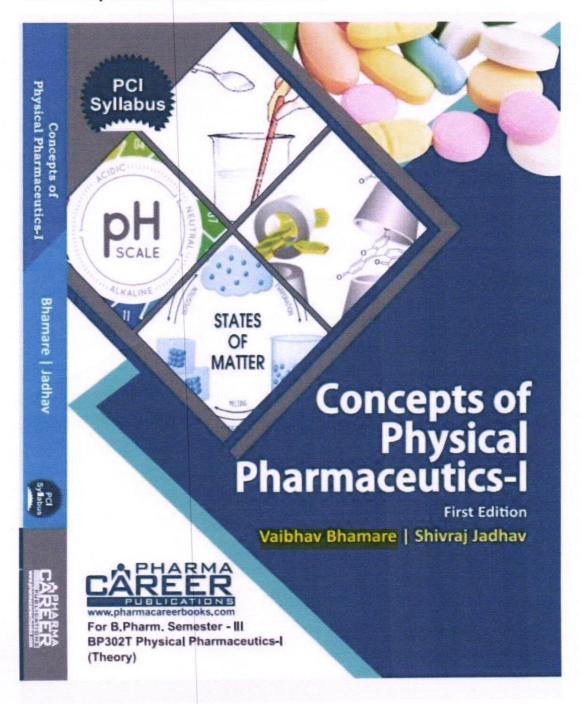
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11	Dr. K. S. Jain	A textbook of Pharmaceutical Organic Chemistry–II	National	2017-2018		K. K. Wagh College of Pharmacy	Nirali Prakashan

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Name of the faculty- Dr. Vaibhav G. Bhamre, Mr. Shivraj P. Jadhav

Name of the publisher- PharmaCareer Publications





# Concepts of Physical Pharmaceutics - I

As per PCI Regulations For B. Pharm: Semester-III

BP302T PHYSICAL PHARMACEUTICS - I (Theory)

FIRST EDITION

#### Dr. Vaibhav G. Bhamare

M.Pharm (Pharmaceutics), Ph. D., M.B.A.
Associate Professor,
K. K. Wagh College of Pharmacy,
Hirabai Haridas Vidyanagari,
Panchavati, Nashik - 422 003

# Mr. Shivraj P. Jadhav

M. Pharm (Pharmaceutics), Shree Shakti Shaikshanik Sanstha's, Divine College of Pharmacy, Nampur Road, Satana - 423 301





K K Wagh College of Pharmacy Nashik-422 003

# **Concepts of Physical Pharmaceutics - I**

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# **CONTENTS**

Unit	Name of Topic	Page No.
I	Solubility of Drugs	01
II	Part I - State of Matter and Properties of Matter	36
	Part II - Physicochemical Properties of Drug	73
III	Surface and Interfacial Phenomenon	91
IV	Complexation and Protein Binding	125
V	pH, Buffers and Isotonic Solution	146



## UNIT : I

# SOLUBILITY OF DRUGS

Solubility expressions, Mechanisms of solute solvent interactions, Ideal solubility parameters, Solvation & Association, Factors influencing solubility of drugs, Diffusion principles in biological systems, Solubility of gas in liquids, Solubility of liquids in liquids, (Binary solutions, ideal solutions), Raoult's law, Ideal and Real solutions, Partially miscible liquids, Critical solution temperature and applications, Distribution law, its limitations and applications.

# 1.1 INTRODUCTION TO SOLUBILITY:

- Solubility is the important parameter to achieve desired concentration of drugs in the
  systemic circulation that could exert desired physiological response. Any drug to be absorbed
  must be present in the form of an aqueous solution at the site of absorption. Since, water
  remains the solvent of choice for liquid pharmaceutical formulations; poor water soluble
  drugs have slow absorption leading to inadequate bioavailability.
- Solubility can be defined in two ways, in quantitative way and qualitative way.
- Quantitatively solubility is defined as the maximum amount of a substance that will dissolve
  in a given amount of solvent at a specified temperature.
- Qualitatively solubility can be defined as spontaneous interaction between solute and solvent to form homogeneous solution.
- Generally, the solubility of a compound depends on the physical and chemical properties
  of the components (solute and the solvent) and several factors that have impact over
  process of solublization such as temperature, pressure, pH, particle size of solute etc.
- Solubility is usually expressed as grams of solute per liter of solvent.

NASHIK-

- A solute is any substance which can be either solid or liquid or gas dissolved in a solvent.
   Generally, in solution, the amount of solute is always less than solvent. For ex. Sodium Chloride, Sugar.
- A solvent is a substance that dissolves a solute. The solvent is the component of a solution
  that is present in higher concentration. Solvents which are generally in liquid form but can
  also be a solid or gas that are used to dissolve, suspend or extract other materials, usually
  without chemically changing either the solvents or the other materials. For ex. Water,

Ethanol.

Title of the book/chapters published- Growing impact of herbal bioenhancers in pharmaceutical industries

Name of the faculty- Dr. Vaibhav Bhamre, Dr. Rakesh Amrutkar, Mr. Vinod Patil, Dr. Chandrashekhar Upasani

Name of the publisher- De-Gruyter, Garmany

**DE GRUYTER** STEM DRUG DELIVERY **TECHNOLOGY** HERBAL BIOENHANCERS IN PHARMACEUTICALS Edited by Prashant L. Pingale





#### Editor

Dr. Prashant L. Pingale

GES's Sir Dr M S Gosavi College of Pharmaceutical Education and Research B-103 Blue Bells Nashik-422005 India

E-Mail: prashantlpingale@gmail.com

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# **Contents**

List of contributing authors - VII

Namita D. Desai, Niserga D. Sawant, Pratima A. Tatke

Chapter 1

Biopotentiation using herbals: novel approach for poorly bioavailable

drugs —— 1

Surendra Agrawal, Pravina Gurjar, Ayushi Agarwal

Chapter 2

Herbal bioenhancers in microparticulate drug delivery — 27

Amarjitsing Rajput, Satish Mandlik, Shipa Dawre, Deepa Mandlik, Prashant Pingale, Shital Butani

Chapter 3

Herbal bioenhancers in nanoparticulate drug delivery system — 45

Shashikant B. Bagade, Shivanee Vyas, Amit B. Page, Kiran D. Patil

Chapter 4

Role of herbal bioenhancers in tuberculosis and drug delivery thereof — 87

Madhur Kulkarni, Roopal Bhat, Suvarna Ingale, Abhijit Date

Chapter 5

Herbal bioenhancers in cancer drug delivery — 103

Rupali A. Patil, Krutika H. Pardeshi, Harshal P. Chavan, Sunil V. Amrutkar

Chapter 6

Pharmacotherapeutics and pharmacokinetics of herbal bioenhancers —— 149

Vaibhav Bhamare, Rakesh Amrutkar, Vinod Patil, Chandrashekhar Upasani Chapter 7

Growing impact of herbal bioenhancers in pharmaceutical industries — 191

Deepa Mandlik, Satish Mandlik, Amarjitsing Rajput

Chapter 8

Herbal bioenhancers and improvement of the bioavailability of drugs — 211



Vaibhav Bhamare, Rakesh Amrutkar, Vinod Patil, Chandrashekhar Upasani

# Chapter 7 Growing impact of herbal bioenhancers in pharmaceutical industries

Abstract: Ayurvedic expertise has made a significant contribution to drug research in the world, with new methods of identifying active compounds. In comparison to modern medicines, herbal medicines have succeeded to emphasize the world for their use, with advantageous therapeutic effects and fewer adverse effects. However, in vitro/in vivo findings for these herbal drugs or extracts are not so impressively correlative. Poor lipid solubility, improper molecular size, and prolonged therapy of phytoconstituents leads to poor absorption, followed by poor bioavailability and treatment expenses. Herbal bioenhancers are nontherapeutic active phytomolecules that, when co-administered at low doses, improve the bioavailability, bioefficacy, and biological activity of different drugs without having a synergistic impact with medication. Since herbal bioenhancers are healthy, nontoxic, inexpensive, easy to obtain, nonaddictive, pharmacologically inert, and nonallergenic, they are enticing pharmaceutical industries as a valuable and most effective means of bioavailability enhancement. Nowadays, industries are focusing on improvement in pharmacokinetic parameters of potent active pharmaceutical ingredients using various bioenhancing mechanisms that can help through alteration in enzyme activity, phytosomal formulation system, escape protein modifications, effects of cholagogic or choleretic agent, and heat production in the organism. An emphasis is made to tackle multidrug resistance in the treatment of infectious diseases. Bioenhancers significantly contribute to the drug development process, with innovative methods for identifying active compounds, using natural drugs and products. Herbal bioenhancers are becoming increasingly popular as a paradigm-shifting technology for improving the bioavailability and bioefficacy of various classes of drugs, nutraceuticals, and veterinary medicines, thanks to recent advances in drug delivery.

Vaibhav Bhamare, Department of Pharmaceutics, K. K. Wagh College of Pharmacy, Nashik 422006, Maharashtra, India, e-mail: vaibhav.bhamre@gmail.com
Rakesh Amrutkar, K. K. Wagh College of Pharmacy, Nashik 422006, Maharashtra, India
Vinod Patil, SPH College of Pharmacy, Malegaon 423105, Maharashtra, India
Chandrashekhar Upasani, SNJB's SSDJ College of Pharmacy, Chandwad 423101, Maharashtra, India

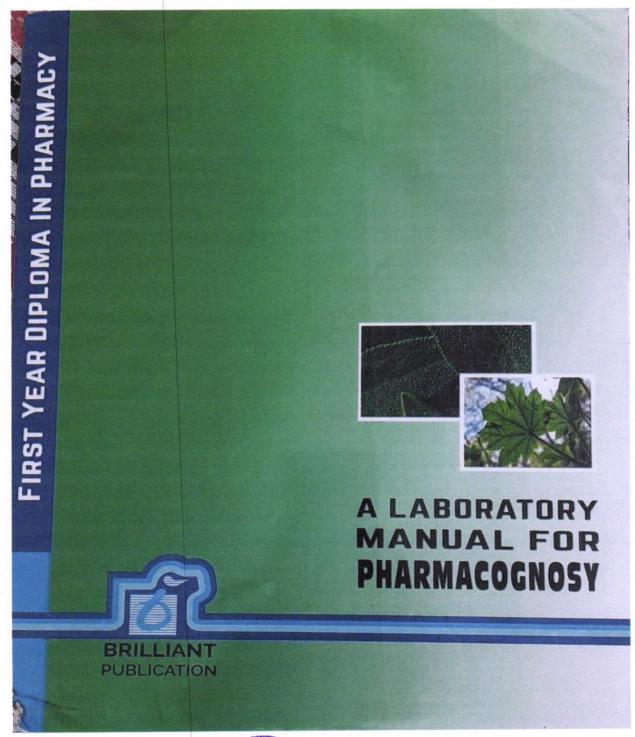
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Name of the publisher- Briliant Publication







A Laboratory Manual is designed as per New Syllabus prescribed by PCI Accordance of E.R 2020

# PHARMACOGNOSY

# FIRST YEAR DIPLOMA IN PHARMACY A LABORATORY MANUAL

#### Salient Features :

- All experiments covered as ER 2020 syllabus.
- Experiments covered in an easy to understand language.
- Learning objectives to aid in understanding the expected set of skills.
- Easy to use format for writing the experiments.
- Concise and Lucid language
- Well labeled diagram of Morphological and microscopically characteristics of crude drug.
- To lend a hand to professionals with scientific interest in herbalism.

- AUTHORS -

Prof. DEEPAK SOMAVANSHI (M. Pharm) Dr. RAJENDRA S. BHAMBAR (M. Pharm, Ph.D., MBA)

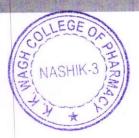
Dr. JITENDRA Y. NEHETE (M. Pharm, Ph.D.) Prof. SHAMAL D. DAWANGE (M. Pharm)

Divine College of Pharmacy, Satana, Nashik - 423301



Puneet Yash Arcade, 3rd Floor, Opp. Kokan Express, Near Kothrud Bus Stand, Kothrud, Pune - 411038. Mob No. - +91 7057488957 / 9156347572 / 8605467166

**Brilliant Publication** 



# **SYLLABUS**

	As per ER 2020, Pharmacognosy Practical Syllabus is as given below:					
Sr. No.	Title					
1.	Morphological Identification of drug:					
	Ispaghula, Senna, Coriander, Fennel, Cardamom, Ginger, Nutmeg, Black Pepper,					
	Cinnamon, Clove, Ephedra, Rauwolfia, Gokhru, Punarnava, Cinchona, Agar.					
2.	Gross anatomical studies (Transverse Section) of the following drugs:					
	Ajwain, Datura, Cinnamon, Cinchona, Coriander, Ashwagandha, Liquorice, Clove,					
	Curcuma, Nuxvomica, Vasaka.					
3.	Physical and chemical tests for evaluation of drugs:					
	Asafoetida, Benzoin, Pale catechu, Black catechu, Castor oil, Acacia, Tragacanth, Agar,					
	Guar gum, Gelatin.					



# **EXPERIMENT NO - 01**

- ♦ Aim To study morphological characteristics of Ispaghula
  - Synonyms Ispaghula, Ispagol, Ishabgula, Spongel seeds.



- **Botanical Source** Ispaghula consists of dried seeds of *Plantago ovata*, belonging to family Plantaginaece.
- Morphology Ispaghula seed -

Colour	Pinkish gray to brown
Odour	None
Taste	Mucilaginous
Shape	Ovate, boat shaped, cymbiform
Size	1.5–3.5 mm long, 1–1.8 mm wide.
Weight of 100 seeds	0.15–0.19 g
Surface	Smooth
Nature	Seeds are hard, translucent
External Features	The dorsal (convex surface) consist of a small elongated reddish brown spot at the center while the ventral (concave surface) has a cavity which is covered with a thin whitish membrane.





# FIRST YEAR B. PHARM | SEMESTER-I

# A PRACTICAL BOOK OF PHARMACEUTICAL ANALYSIS

Dr. K. S. JAIN

Ms. D. K. KADAM









# **A Practical Book of**

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As Per PCI Regulations

# FIRST YEAR B. PHARMACY Semester - I

Dr. K. S. Jain

M. Pharm., Ph.D., FIC
Principal & Profesor,
Deptt. of Pharmaceutical Chemistry,
K.K.Wagh Education Society's,
K.K.Wagh College of Pharmacy,
Hirabai Haridas Vidyanagari, Amrut Dham, Panchavati,
Nashik, 422003.

Ms. D. K. Kadam

M. Pharm.,
Asst. Professor,
Deptt. of Pharmaceutical Chemistry,
K.K.Wagh College of Pharmacy,
Nashik, 422003.

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(For orders within Pune)

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Email: niralidelhi@pragationline.com

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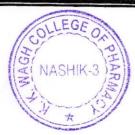
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# Contents

- Luction	1.10
Unit 1 : General Introduction	1,1
1.1 Volumetric Glassware	1.1
1.1.1 Pipettes	1.5
1.1.2 Burettes 1.1.3 Measuring Cylinders	1.6
· · · · · · · · · · · · · · · · · · ·	1;
	1;
<ul><li>1.1.5 Erlenmeyer Flasks</li><li>1.1.6 Weighing Bottles and Specific Gravity Bottles</li></ul>	1,
es la production de la constant de l	1,9
1.1.7 Beakers	1.9
1.2 Cleaning of Glassware	moissans 19
1.3 Some Basic Concepts	
1.3.1 Preparation of Aqueous Solutions	19
1.3.2 Per cent Solutions	1.11
1.4 Volumetric Methods	1.11
1.4.1 Definitions of Terms	1.11
1.4.2 Requirements for Volumetric Methods	1.12
1.4.3 Classification of Volumetric Methods	1.12
1.4.4 Steps Involved in Quantitative Analysis	1.13
1.4.5 End Point Detection	1.14
1.4.6 Calibration of Volumetric Glasswares	1.15
Question Bank for Viva-Voce	1.16
register and one	
Unit 2 : Limits Tests	2.1 - 2.20
Experiment 1: To Perform Limit Test for Chlorides	2.3
Experiment 2 : To Perform Limit Test for Sulphate	2.6
Experiment 3: To Perform Limit Test for Iron	2.0
Experiment 4: To Perform Limit Test for Arsenic	
Question Bank for Viva-Voce	2.11
The state of the s	2.17
Init 3 Preparation and Standardination of S	: 10.076
Init 3 Preparation and Standardization of Some Volumetric Reagent	Solutions
	3.1 - 3.14
Experiment 1: To Prepare and Standardize 0.1 N Sodium Hydroxide	3.1
The repair and Standardize of M.C. I.	3.3
· To riepale and Standardize 0.1 N.C. II	
TO THE MILL STANDARDING OF A LA	3.00
	ite 5.0
Question Bank for Viva-Voce	hate 3.10
and the second s	3.11



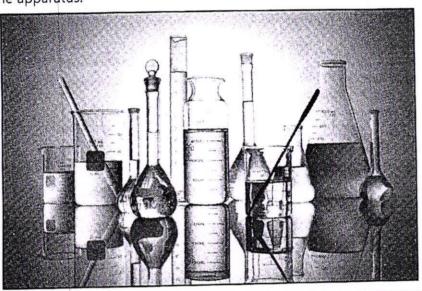
	it 4: Assay of the Compounds Along with Standardization of Titrant	4.1 - 4.14
Un	Experiment 1 : Assay of Ammonium Chloride by Acid-Base Titration	4.1
	Experiment 1 : Assay of Ammonium Chiolide by Acid Base Manager	4.2
	Experiment 2: Assay of Ferrous Sulphate by Cerimetry	4.3
	Experiment 3: Assay of Copper Sulphate by Iodometry	4.5
	Experiment 4: Assay of Calcium Gluconate by Complexometry	
	Experiment 5: Assay of Hydrogen Peroxide by Permanganometry	4.7
	Experiment 6: Assay of Sodium Benzoate by Non-Aqueous Titration	4.9
	Experiment 7: Assay of Sodium Chloride by Precipitation Titration	4.10
	Question Bank for Viva-Voce	4.12
	Question built io	
	it 5 : Determination of Normality by Electro-Analytical Methods	5.1 - 5.10
		5.1
1.	Conductometric Titrations	Hydrochloric
	Experiment 1: Determination of the Strength of a Solution of Strong Acid (	xide) 5.1
	Acid) by A Standard Solution of Strong Base (Sodium Hydro:	: da (0.1 N L) Cl
	Experiment 2: To Carry Out the Conductometric Titration of Mixture of Ac	5.5
	+ 0.N CH₃COOH] vs. Strong Base (1 N NaOH)	14000
2.	Potentiometric Titrations	5.7
	Experiment 3: Determination of Strength of Strong Acid Against Strong	ong Base by
	Potentiometric Titration	5.7
	Question Bank for Viva-Voce	5.9
	Question builties in a	
n:	bliography	B.1 - B.1
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# **GENERAL INTRODCUTION**

All laboratory work involves some form of measurement. Various types of volumetric glassware are available for measuring lab reagents. Before starting any analytical experiment, one must familiarize him/herself with the types of volumetric glassware and should learn to use the apparatus accurately to eliminate the errors resulting from careless handling of the apparatus.



# 1.1 VOLUMETRIC GLASSWARE

#### 1.1.1 Pipettes

These are used to transfer a known volume of liquid from one container to another. They are designed either to contain (TC) a specific volume of liquid or to deliver (TD) a specified volume. [To contain and To deliver (TC & TD): Manufacturer stamp TC or TD near the top of the pipette. TC pipette holds or contains a particular volume but does not dispense that exact volume, whereas TD pipette will dispense the exact volume indicated].

Table 1.1 : Pipette Classification

- I. Design
  - A. To contain (TC)
  - B. To deliver (TD)

(1.1)







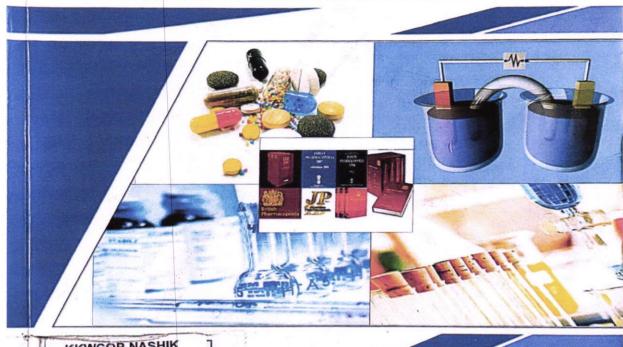
FIRST YEAR B. PHARM | SEMESTER-I

# PHARMACEUTICAL ANALYSIS

Dr. K. S. JAIN

Ms. D. K. KADAM

Mrs. K. P. BAVISKAR



KKWCOP NASHIK 615,19/JAI B3890







# PHARMACEUTICAL ANALYSIS

As per PCI Regulations

First Year B. Pharm. Semester - I [BP102T]

#### Dr. K. S. Jain

M. Pharm., Ph.D., FIC
Principal & Professor
Deptt. of Pharmaceutical Chemistry
K. K. Wagh Education Society's
K. K. Wagh College of Pharmacy
Hirabai Haridas Vidyanagari, Amrut Dham, Panchavati
Nashik 422003

#### Ms. D. K. Kadam

M. Pharm.
Asst. Professor
Deptt. of Pharmaceutical Chemistry
K. K. Wagh College of Pharmacy
Nashik 422003

# Mrs. K. P. Baviskar

M. Pharm,
Asst. Professor
Deptt. of Pharmaceutical Chemistry
K. K. Wagh College of Pharmacy
Nashik 422003



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Pharmaceutical Analysis

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# Contents

Unit I : Introduction to Pharmaceutical Analysis	B-09 6
	1.1 - 1.18
<ol> <li>Pharmaceutical Analysis</li> <li>1.1 Introduction: Definition &amp; Scope of Pharmaceutical Analysis</li> </ol>	1.1
1.1 Introduction: Definition & Scope of Frialmass	1.2
1.2 Different Techniques of Analysis	1.4
1.3 Methods of Expressing Concentration	1.6
1.4 Primary and Secondary Standards	1.6
1.4.1 Primary Standards	1.8
<ul> <li>1.4.2 Secondary Standards</li> <li>1.5 Preparation and Standardization of Various Molar and Normal Solutions:</li> </ul>	d Fhar
Oxalic Acid, Sodium Hydroxide, Hydrochloric Acid, Sodium Thiosulphate,	3 Thet
Sulphuric Acid, Potassium Permanganate and Ceric Ammonium Sulphate	1.9
1.5.1 Preparation of Standard Solutions	1.9
The state of the s	7.0
1.5.1.1 Preparation of N/10 H <sub>2</sub> SO <sub>4</sub> 1.5.1.2 Preparation of N/10 Na <sub>2</sub> CO <sub>3</sub> Solution	
1.5.1.3 Preparation of N/10 NaOH Solution	1.12
2 St. 30 Sp. 10	1.12
	1.12
1.5.1.5 Preparation of N/10 KMnO <sub>4</sub> Solution 1.5.1.6 Preparation of 0.1 N Hydrochloric Acid	1.13
1.5.1.7 Sodium Thiosulfate, (0.1 N) Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> · 5H <sub>2</sub> O (248.19)	1.14
1.5.1.8 Cerric Ammonium Sulfate, (0.1 M) (NH <sub>4</sub> ) <sub>4</sub> Ce(SO <sub>4</sub> ) <sub>4</sub> 2 H <sub>2</sub> O	1.15
Question Bank	75.68
2. Errors in Chemical Analysis 2.1 Introduction and assessment of the second of the se	.1 - 2.10
2.2 Precision	
t 2.3 Accuracy	2.2
and the property of the contract of the contra	2.3
	2.3
and the state of t	2.4
	2.4
2.7 Methods of Minimizing Errors	27
2.8 Methods to Minimize and Detect Determinate Errors	27
Question Bank	
3. Pharmacopoeia, Sources of Impurities, Limit Tests 3.1 Pharmacopoeia 3.	1 - 3.40
The macopoeia	3.1
3.1.1 Indian Pharmacopoeia	2.1
3.1.2 British Pharmacopoeia	
3.1.3 European Pharmacopoeia	
·	3.6



3.1.4 United States Pharmacopoeia (USP)	3.7
3.1.5 Pharmacopoeia International (International Pharmacopoeia)	3.8
3.1.6 Extrapharmacopoeia (Martindale)	3.9
3.2 Sources of Impurities in Pharmaceuticals	3.10
3.2.1 Sources and Effects of Impurities in Pharmacopoeial Substances	3.10
3.2.2 Effects of Impurities on Pharmaceuticals	3.13
3,3 Limit Tests	3.13
3.3.1 Importance of Limit Tests	3.13
3.3.2 Principle and Procedure of Limit Test for Chloride	3.15
3.3.2.1 Principle and Procedure of Modified Limit Test for Chloride	3.16
3.3.2.2 Principle and Procedure of Limit Test for Sulphate	3.17
3.3.2.3 Principle and Procedure of Limit Test for Sulphate	3.19
3.3.2.4 Principle and Procedure of Limit Test for Iron	3.20
3.3.2.5 Principle and Procedure of Limit Test for Lead	3.22
3.3.2.6 Principle and Procedure of Limit Test for Arsenic	3.27
3.3.2.7 Principle and Modified Procedure of Limit Test for Arseni	
Different Compounds	3.30
3.3.2.8 Principle and Procedure of Limit Test for Heavy Metals	3.33
C.Z. Jonan Pea Question Bank pentagon in the related by 1.0 to not success 1.5.0.5	3.37
E.7 ethome of Unit II FU to no manage in C.S.D.2	
4. Acid Base Titrations Shipping and Automotive E.C. 4.1 -	4.20
4.1 Concepts of Acid and Base	4.1
411 Arrhenius Theory III 90009000 one 9000000 murpok to abbaselika	4.1
4.1.2 Lowry and Bronsted's Theory	4.1
4.1.3 Lewis's Theory	4.3
4.1.4 Usanovich Theory	4.3
415 Lux Flood Concept	4.4
4.2 Theories of Acid-Base Indicators	4.4
4.2.1 Indicator Range	4.5
128300 noticed	4.5
4.2.2 Ostward's Theory  4.2.3 Quinonoid Theory	4.7
4.3 Theory Involved in Titrations of Strong, Weak and Very Weak Acids and Base	
4.4 Buffer Equations and Buffer Capacity	4.8
4.4 Bullet Equations and Bullet Capacity	4.11
4.5 Neutralization curves	4.11
4.5.1 Neutralization curve for Strong Acid and Strong Base Titudion	4.13
4.5.2 Predicting date for Fredrick Acid and Scholing base Title	
4.5.3 Titration Curve between Weak Base and Strong Acid Titration	4.15
4.5.4 Titration Curve between Weak Acid and Weak Base	4.17
Question Bank	4.19





	Control Street South	5.1 - 5.1
. /	Ion-Aqueous Titrations	5.:
V :		5.:
	2 Advantages of Non-aqueous Titrations	5.2
. 5		5.2
	5.3.1 Properties of Non-aqueous Solvent	5.2
811	5.3.2 Different Types of Solvents	5.4
5	3	5.4
5	<b>3</b>	5.4
81.5	5.5.1 Preparation of 0.1 N Perchloric Acid	5.5
315	5.5.2 Standardization of 0.1 N Perchloric Acid	5.6
	5.5.3 Choice of Indicators	
611	5.5.4 Effect of Temperature on Assays	5.6
5.		5.7
***	5.6.1 Acidimetry in Non-aqueous Titrations	5.7
	5.6.1.1 Titration of Primary, Secondary and Tertiary Amines	5.7
J.E	5.6.1.2 Titration of Halogen Acid Salts of Bases	5.8
2.2	5.6.1.2.1 Amitriptyline Hydrochloride	5.8
	5.6.2 Alkalimetry in Non-Aqueous Titrations	5.8
	5.6.2.1 Preparation of 0.1 N Potassium Methoxide in Toluene-M	ethanol 5.9
	5.6.2.2 Preparation of 0.1 N Sodium Methoxide	5.9
- 1.20	5.6.2.3 Preparation of 0.1 N Lithium Methoxide	5.9
5.7	5.6.2.4 Standardization of 0.1 N Methoxide Solution	5.9
5.8	Estimation of Sodium Benzoate and Ephedrine HCI	5.10
٥.د	Applications of Non-aqueous litration	5.11
100	Question Bank	5.12
	Unit III	
6. Pre	ipitation Titrations	61 610
6.1	Introduction	
6.2	Solution Process	6.1
6.3	Factors Affecting Solubility	6.1
6.4	Detection of End Point	6.2
	6.4.1 Mohr's Method	6.5
	6.4.2 Volhard's Method	
	6.4.3 Fajan's Method	6.6
6.5	Estimation of Sodium Chloride	
6.6	Classification of Methods in Precipitation Titration	
6.7	Applications of Precipitation Titrations  • Question Bank	6.9
	Question Bank	6.10
	- 100 may 200 -	6.10





7. Com	plexometric	Titrations			7.1	- 7.16
7.1	Introduction	and Definition				7.1
9 7.2	Ligands (Cor	nplexing Agent)				7.2
8.F	7.2.1 Classif	fication of Ligands				7.2
8.7	7.2.2 Chelat	ting Agents				7.2
2.7	7.2.3 EDTA		to tal			7.3
₹ 7.3	Stability Con	stant				7.4
3.8		rs Affecting Stability				7.5
8.8	7.3.1.1					7.5
2.8	7.3.1.2	? Thermodynamic/	Chelon Effect	a ward to		7.6
8,8	7.3.1.3					7.6
8.8	7.3.1.4					7.6
8.5	7.3.1.5					7.6
7.4	Detection of	End Point	•			7.7
12.8	7.4.1 Visual	Indicators	Hakriga ONg e	Aparteoperal E	Ftg	7.7
ILS	7.4.1.1					7.7
118	7.4.1.2			ons in Grayanası		7.10
81.6	7.4.2 Instru	mental Methods	. 93	or Sadum Sulpha	. สอใหมาใชร	7.10
8.15		Potentiometric M			8 10.1 9:-	7.10
8115	7.4.2.2	2 Spectrophotome	tric Method	976000	1010 St. S	7.10
8.16	7.4.2.3	Amperometric M	ethod	, Bank .	o tega O - +	7.10
7.5	Types of Titr	ation (no	Last emile	sneibo2) noi 'm	ill color is	7.10
180	7.5.1 Direct	Titrations			romoutogeni.	7.10
LÉ		Titration			yro, a	7.11 7.11
9.2	7.5.3 Replac	cement Titration/ D	isplacement T	itration	or disco.	7.11
9.2	7.5.4 Indire			of Thatian	SAL Pater	7.12
		metric Titration		010/519	9.3.2 Temp	
2.6 2.6 2.7.7	Titration of N	of Complexometric	Titrations		elujonis	
5.0 7.8	Applications	f Magnesium Sulph	ate 200	ใหลายไปการเสรา เรื่อ	aid to ear T	7.14
£ 6 7.9		f Calcium Gluconate	\$ 100 mm		Band of	
	Question			hodtaM bi	ing (see	7.15
€.0 3. ₀ Grav		1 Dank		indefine	iber 4 8	.1 - 8.18
8.1	Introduction	water South	AZ AZ I PONEÎ	estionabnedé Lus		8.1
8.2	Types of Pre	cipitates	graph and soul	ation of LM Scu	Maria 13.6	
E. 8.2	8.21 Crysta		MATE AND ASSESSED	Malio aphasilis	Arrest Carl	8.2 8.3
8.8	8.2.2 Colloi	dal Precipitates	a local mar DE	other plants	1427-1 - 15.848	8.3 8.4
9.4	8.2.2.1		al Precipitates	A mpsecimen.		8.4
₹.0	8.2.2.	2 Peptization of Co	olloids	Wednesia, man		8.5
9.5	8.2.3 Gelati	inous Precipitates				-



			Durance and Da	rticle Size		STATE NAME	ALM WHILL	8.5
8.	Preci	pitation	Process and Pa	twald Ripen	ning)	C that action	Forms	8.6
8.4	<ul> <li>8.3 Precipitation Process and Particle Size</li> <li>8.4 Digestion of Precipitates (Ostwald Rip</li> <li>8.5 Purity of Precipitates</li> </ul>							8.6
8.:						- 1751 p. 1. 128 l I	1 1 2 5	8.8
	8.5.1		cipitation Surface Adsor	rntion	7 1/18	e e almi	227	8.
2.7						6.731	67.	8.
		8.5.1.2	Mixed Crystal	Machanica	l Entrapment	toote (1) f	ni enie	8.
. a.	0.5.0		Occiusion and	1 Mechanica	- 100 c 1000	and softs.	4.5.5	8.8
0.6	8.5.Z	POST Pre	ecipitation	ous Solution	n doine	offic Land		8.8
8.6			rom Homogen	003 30101101	y Jenna A	T CLES		8.9
8.7		oitating A	and the same of th		residents			8.9
			ic Precipitants					8.9
3.1			g Agents		ng at garanta an pagasal Perupaka			8.9
4.	8.7.3							8.1
					ecipitants			8.11
					Precipitants			8.11
8.8					i Dominala alto.			8.11
8.9					, ** (F.P.C.) \$7.1			
8.10					i Methods			8.15
		Principl			en kuomaline			8.15
		Procedi			h metarigados			8.15
		estion B			- Chite Horse			8.16
			on (Sodium N	litrite Titra	tion)	prosecuti sz Beser TrospO		
9.1	Introdu		34	20		rau i nerau Back Nicasa		9.1
9.2	Theory		rode: T	mamando	en. 10 Tenauery Da			9.1
9.3			iazotization			vil treapni		9.2
<u></u>		Rate of T				alter Planta		9.2
11.7	9.3.2	Tempera:	ture			usai to no	4	9.2
9.4	Princip	le		an oligini?	sinze molte/env			9.2
9.5	Types o	of Diazot	ization Titration		sique meter.			9.3
1.5	9.5.1	Direct Tit	rations		sdenogradi Arre			9.3
dir	9.5.2 F	Reverse N	/lethod			ero i noltzek	0. %	9.3
	9.5.3 S	pecial M	ethod			A	y a comia	
9.6	Prepara	tion and	Standardization	on of 1 M So	odium Nitrate	- strifting	aoni d	9.3
			on of 1M Sodiu		461	ci importa	EXHIPT I	9.3
1.0	9.6.2 S	tandardi	zation of 1M S	odium Nitra	ate with Sulpha	nilamida	LS.8	9.3
9.7	Assay o	f Calciun	n Aminosalicyla	ate	- Turn Sulpris	imamide	73	
			Diazotization T		- 10.7 Mass	1 1143		9.4
, <b>9</b>		stion Ba		W.	object mast,	1. L. S. A. S.	189	9.5
	4		/ IE	ie i	E. 1. 20	a collabo	~ 3	9.5
			Alm Lake	ENI		- / /	1 1	





			Unit IV			
0. Redo	x Titrat	ions		10.1 - 10.16		
	Introduction					
10.2	Concepts of Oxidation and Reduction					
10.3	Balancing of Half Reactions					
	Equivalent Weights					
	Redox Potential					
10.6	Detection of End Point in Redox Titrations					
1.7.5		Internal In				
1,52		Self Indica	ENC. 2	10.6 10.6		
1.31.		External In	· ************************************	10.0		
121		Specific Ir	The state of the s	10.0		
B. San		0. 0.000	netric Method			
10.7		Titrations	at of D Seember, 8-d Points			
	10.7.1	Titrations	using Potassium Permanganate	10.7		
177	10.7.2	Cerimetry				
	10.7.3	Iodine Tit				
M.E.I -	10.7.4	Bromator		10.13		
	10.7.5	Dichrome	etry	10.14		
	10.7.6	Titration v	with Potassium Iodate			
LEI	• Que	estion Ban		n 19 19		
2.64		Unit V	: Electrochemical Methods of Analysis	196. T		
)	J.,	2012	Chair shapers and property of the second shapers in 2 Pts invan Steeting as	11.1 - 11.13		
1.0	ductometry		95 CHILAIS (MASH N. 1)	11.1		
11.1	Introdu	iction	Oregoing Mercley Endrode	11.4		
11.2	Measur	rement of	Conductivity	11.7		
11.3	Conduc	ctometric	Titrations • Titrations	11.7		
ILI	11.3.1		C IIII dilono	110		
1.7			Strong Acid with a strong base	11.8		
5.1	X.	11.3.1.2	Strong Acid with Weak Base	11.9		
P 52 . 4	e de		Weak Acid with a Strong Base	11.9		
1.102		11.3.1.4	Weak Acid with Weak Base	11.10		
	11.3.2	Precipitat	tion Titrations	11.10		
		Redox Tit		11.11		
	1134	Complex	ometric Titrations	11.11		
	44.0.1					
	11 3 5	Non-aqu	eous Titrations	11.12		
11 /	11.3.5	Non-aqu	eous Titrations	11.12		
11.4	11.3.5 Applica	Non-aqu	eous Titrations	11.12 11.12		





12. Potentiometry	127		12	.1 - 12.1
12.1 Introduction			dani.	корон 12
12.2 Electrode Potential			ADO IN	12
12.3 Cell		as The net, pint to st	390/10)	12.
12.3.1 Galvanic Cell		e treating go hi		
12.4 Electrodes		James Mary		
12.4.1 Reference Electro		ie . 910		
12.4.1.1 The Sta		Electrode (SHE)		
		μ_jc 5 . 1		
12.4.1.3 Silver Cl	hloride Electrode	\$10°, 31, 1, 1,95	3.0.01	12.7
12.4.2 Indicator Electrod	es	External is alx		12.8
12.4.2.1 Metal El	ectrode	Specific Indicators	40.01	12.8
12.4.2.2 Glass Ele	ectrode	Para in groeter Mained		12 0
12.5 Methods to Determine En		ខាងសម្រា	26.0	12.10
12.6 Types of Potentiometric Ti		erado Espire en arguni	17.01	12.11
12.7 Applications of Potentiom	etric Titrations	Cemmeir	10.72	
<ul> <li>Question Bank</li> </ul>	(4	างสะการ ขณะเล้	Erm	
13. Polarography		V* Filther and	13.1	
13.1 Introduction		v (Secretaria)		13.1
13.2 Principle		ugation var. PC 1000		
13.3 Theory of Polarography				_
13.3.1 Polarization of Elect	rodes	he3 ngung		13.2
13.4 Construction and Working of	of Dropping Mer	Curv Claster de la		13.2
Rotating Platinum Electrode	or propping Mei	cury electrode and		9
13.4.1 Dropping Mercury El				13.3
13.4.2 Rotating Platinum Ele	ectrode	resident		13.3
13.4.3 Polarographic Ad L	ectrode	Vivita i 100 to memore	iene M.	13.8
13.4.3 Polarographic Modes 13.5 Applications	S	WHO SHE SHEET OF	1.403	8 120
		Pagaran Magaran Mila		13.11
Question Bank		MARGINE LINE		13.13
Index	the state of a	DAC SPRINGE		
Bibliography	all inext rain	List sees Ald		1 - 1.2
	120 mm = 1	minimal management	В.	1 - B.1



# Unit I: Introduction to Pharmaceutical Analysis

# Chapter 1

# PHARMACEUTICAL ANALYSIS

### LEARNING OBJECTIVES +

After completing this chapter, student should be able to understand:

- Definition and scope of Pharmaceutical anlaysis.
- Various analytical techniques.
- Methods of expressing concentration.
- Standard solutions used in analysis.
- Preparation of solutions of various concentrations.

# 1.1 INTRODUCTION: DEFINITION & SCOPE OF PHARMACEUTICAL ANALYSIS

#### **Definition:**

**Pharmaceutical analysis** is a branch of practical chemistry that involves a series of processes for identification, determination, quantification and purification of a substance, separation of the components of a solution or mixture, or determination of structure of chemical compounds. The substance may be a single compound or a mixture of compounds and it may be isolated or in any of the dosage forms. The substances used as pharmaceuticals are from various synthetic or natural (animal, plant, marine, microbial or mineral) sources.

#### Scope:

The process of analysis can be broadly categorized as; a) qualitative (identification) and b) quantitative (estimation). The sample to be analysed is called as **analyte** and on the basis of size of analyte, quantitative analysis can be termed as; macro (0.1 gm or more), semimicro (0.01 gm to 0.1 gm), micro (0.001 gm to 0.01 gm), sub-micro (0.0001 gm to 0.001 gm), ultra-micro (below 10<sup>-4</sup> gm) and trace analysis (100 to 10000 ppm).

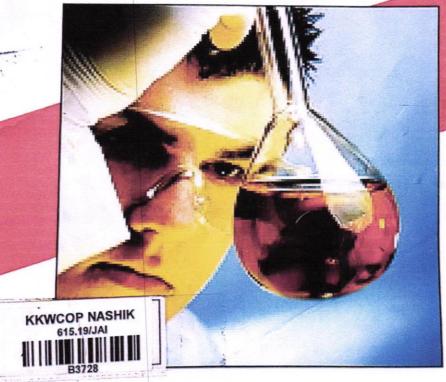
- Qualitative analysis is performed to establish composition of natural/synthetic substances. These tests are performed to indicate whether the substance or compound is present in the sample or not. Various qualitative tests involve; detection of evolved gas, formation of precipitates, limit tests, colour change reactions, determination of melting point and boiling point etc.
- 2. Quantitative analysis is mainly used to quantify any compound or substance in the sample. These techniques are based on (a) the quantitative performance of suitable chemical reaction and either measuring the amount of reagent added to complete



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Nashik 422003

Dr. P. B. Miniyar

M. Pharm., Ph.D., FAGE
Professor and Head
Sinhgad Technical Education Society's
Sinhgad Institute of Pharmacy, Narhe
Pune 411 041

#### Dr. L. V. G. Nargund

M. Pharm., Ph.D., FIC
Principal & Professor,
Deptt. of Pharmaceutical Chemistry
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## Contents

1.	Systematic C	lu	alitative Analysis of	
			nic Compounds	1.
	1. Preliminary			1.1 - 1.82
	2. Elemental A	na	lysis or Lassaigne's Test	1.6
	3. Compounds	: E	lements wise and Functional groupwise	1.17
	4. Preparation	of	Derivatives	1.21
			ants and Derivatives	1.54
			Binary Mixture	1.67
2.			ular Models of Simple Covalent Molecules	1.78
	Laboratory T			2.1 - 2.8
	The state of the s			3.1 - 3.11
4.	Analysis of C			4.1 - 4.7
	Experiment 1		Determination of Acid Value	4.2
	Experiment 2		Determination of Saponifaction Value	4.3
	Experiment 3	:	Determination of Ester Value	4.4
	Experiment 4	:	Determination of Peroxide Value	4.4
_	Experiment 5	:	Determination of Iodine Value	4.5
5.	Preparation	of		5.1 - 5.30
	Experiment 1	:	Synthesis of Benzanilide from Aniline	5.1
	Experiment 2	:	Synthesis of 2 Naphthyl Benzoate from β-Naphthol	5.2
	Experiment 3	:	Synthesis of Phenyl Benzoate from Phenol	5.4
	Experiment 4	:	Synthesis of Acetanilide from Aniline	5.6
	Experiment 5	:	Synthesis of p-nitroacetanilide from Acetonilide	5.8
ď.	Experiment 6	:	Synthesis of 2, 4, 6-tribromoaniline from Aniline	5.11
1	4Experiment 7	:	Synthesis of p-bromoacetanilide from Acetanilide	5.13
	Experiment 8	:	Synthesis of m-dinitrobenzene from NItrobenzene	5.16
	Experiment 9	:	Synthesis of Nitrobenzene by Nitration Reaction	5.18
	Experiment 10	:	Synthesis of Benzoic acid from Benzyl chloride by Oxidation Re	
- 1	3Experiment 11	:	Synthesis of Benzoic Acid from ethyl Benzoate by base Hydrol	ysis 5.20
	Experiment 12	:	Synthesis of Salicyclic Acid from Alkyl Salicylate by	r 22
			Hydrolysis Reactions	5.22
	Experiment 13	:	Synthesis of Benzil from Benzoin	5.24
	Experiment 14	:	Synthesis of Phenylazo-2-naphthol from Aniline by	5.25
			Diazotisation Coupling reaction	5.26
	Experiment 15	:	Synthesis of Dibenzalacetone from Benzaldehyde	5.28
1	Experiment 16	:	Synthesis of Cinnamic Acid from Benzaldehyde	5,22





## Systematic Qualitative Analysis of Unknown Organic Compounds

Qualitative organic analysis is the systematic identification and confirmation of an unknown organic compound.

Qualitative organic analysis forms an important part of the practical training of the budding organic chemist. It gives a balance between structured systemic approach and the intuition, which the student acquires with experience.

Proper application of knowledge of organic chemistry and significant observations can help draw meaningful conclusions leading to pinpoint identification.

The systematic steps of qualitative analysis:

Following essential and integral steps for the systematic identification of an unknown organic compound, should be followed meticulously and religiously, skipping none of it.

1. Preliminary tests: It can tell the state, nature and type in a broad manner.

2. Detection of elements/elemental analysis (Lassaigne's test): It can throw light on which of the elements N, [O], S, halogens are present.

3. Detection of functional group: Which is the main organic functional group in the compound.

4. Derivative preparation: This is preparable and can help to confirm the unknown compound.

5. Determination of physical constants: B.P. or M.P. are an important characteristic physical property.

6. Separation of binary mixture: In case of mixture of two unknown compounds, which need to be separated prior to their analysis.

This helps in error free identification of an unknown organic compound. The tests are given in sequence one after the other, to reach the end results and to confirm the identity of the given, unknown organic compound, correctly and student is required to follow the same sequence.

Main class of organic compounds encountered in the systematic qualitative analysis: (a) Carbohydrates: (e.g., glucose, fructose, súcrose, lactose, maltose, starch, dextrin, etc.) Tests:

Molisch test: General test for carbohydrates.

Barfoed's test: For distinction between mono and disaccharides.

3. Fehling's and Benedict's tests: For differentiating between a reducing and a non-reducing saccharides.

4. Selvinoff's test: Test for ketoses.

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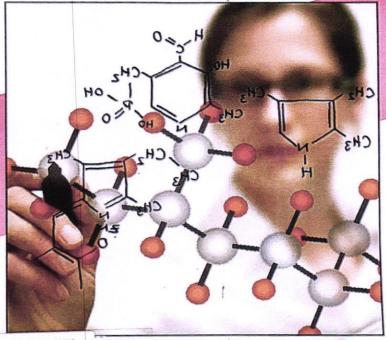


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Principal & Professor, Deptt. of Pharmaceutical Chemistry
K.K. Wagh Education Society's
K.K. Wagh College of Pharmacy
Hirabai Haridas Vidyanagari,
Amrut Dham, Panchavati, Nashik 422003

Dr. P. B. Miniyar

M. Pharm., Ph.D., FAGE
Professor and Vice-Principal
Sinhgad Technical Education Society's
Sinhgad Institute of Pharmacy, Narhe
Pune 411 041

#### Dr. L. V. G. Nargund

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Principal and Professor
Deptt. of Pharmaceutical Chemistry
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## Contents

#### Unit I

	. c	lassific	ation.	Nomenclature and Isomerism	1.1 - 1.37
		1 Class	ification o	of Organic Compounds	1.
		2 Com	mon nam	es of Organic Compounds	1.4
	,	121	Compoi	unds containing Carbon and Hydrogen Uniy	1,4
		122	Compoi	unds containing Carbon, Hydrogen and Oxygen	1,0
	1	1.2.3	Compou	unds containing Carbon, Hydrogen and Nitrogen only	1.9
		124	Compou	unds containing Carbon, Hydrogen, Oxygen and Halogunds containing Carbon, Hydrogen, Oxygen, Nitrogen	ens only 1.10
			and Hale	ogens only	1.13
			Sulphur	unds containing Carbon, Hydrogen, Oxygen, Nitrogen, and/or Halogens	1.13
	1.3	IUPA(	Nomen	clature System for Organic Compounds: General Rules	1.15
	1.4	Isome	erism		1.27
		1.4.1	Constitu	tional or Structural Isomerism	1.27
			Stereoise	omerism	1.32
-		<ul> <li>Q</li> </ul>	uestions		1.34
_				Unit II	
2.		kanes	- 37		2.1 - 2.8
	2.1				2.1
	2.2	,		f Atomic Orbitals	2.3
				f Alkanes	2.5
	2.4			s (alkanes)	2.7
			estions		2.8
3.		enes			3.1 - 3.18
	3.1	Introd	uction: Pr	operties of Alkenes	3.1
	3.2			ctions Type	3.4
		3.2.1	The Med	hanisms of Elimination Reaction (β-Elimination)	3.5
		3.2.2	Orientati	ion of Elimination	3.7
		3.2.3		son of E <sub>1</sub> , E <sub>2</sub> and E <sub>1cB</sub> Reactions	3.12
	2.2	3.2.4	Ozonoly	sis	3.13
	3.3	Additio	n Reactio	ons of Alkenes: Electrophilic Additions	3.14
	3.4	The Ma	irkovniko	ff's Orientation in Electrophilic Addition	3.17
	_	• Qui	estions		3.18
١.			d Diene	\$	4.1 - 4.9
	4.1	Introdu			4.1
	4.2	Stability	of Conju	gated Dienes	4.2
	4.3	Diel's-A	lder Read	tion	4.4
	4.4	Electrop	hilic Add	lition Reaction	4.5
	4.5	Free Ra	dical Add	ition Reactions	4.7
	4.6	Allylic R	earrange	ment	4.8
			stions		4.9
					4.5





#### Unit III

5.	AI	kyl Ha	lidae	
Э.	5.1	-	duction	5.1 - 5.24
	5.2		eral Reactions of Alkyl Halides	5.1
	5.3	Facto	ors affecting SN <sub>1</sub> and SN <sub>2</sub> Reaction Mechanisms	5.4
,	5.4	Subs	titution Vs Elimination	5.11
	5.5		tures and Uses of Some Alkyl halide derivatives	5.14
	4, 1		Questions	5.17
6.	Alc	ohols		5.23 <b>6.1 – 6.10</b>
	6.1	Intro	duction	6.1
	6.2	Nome	enclature	6.2
	6.3	Physic	cal Properties	6.3
	6.4	-	tative Tests	6.4
	6.5	Struct	ture and Uses of some Alcohols	6.6
	19 38	• Q	uestions	6.10
			Unit IV	
	_		• • • • • • • • • • • • • • • • • • • •	
7.			Compounds	7.1 – 7.28
	7.1		ophilic Addition	7.1
			Mechanism of Nucleophilic Addition and Condensatio	
		7.1.2	Examples of Nucleophilic Addition to Carbonyl Groups	
		7.1.3	Acetal-Preparation and Uses / Addition of Alcohols to	
		7.1.4		7.6
	7.2		omeric Effect	7.12
	7.3		Reactions of Carbonyl Compounds	7.13
		7.3.1	Aldol Condensation	7.13
		7.3.2	Crossed Aldol Condensation	7.14
.~		7.3.3	Cannizaro Reaction	7.14
		7.3.4	Crossed Cannizaro Reaction	7.15
		7.3.5	Benzoin Condensation	7.15
		7.3.6	Perkin Reaction	7.16
	7.4		ative tests for Aldehydes and Ketones	7.17
		7.4.1	2, 4-DNP Test for Aldehydes and Ketones	7.17
		7.4.2	Tollen's Test for Aldehydes	7.17
		7.4.3	Jones (Chromic Acid) Oxidation Test for Aldehydes	7.18
		7.4.4	Iodoform Test for Methyl Ketones	7.19
	7.5	Structu	res and Uses of some Carbonyl Compounds	7.19
		• Qu	estions	7.28
	35235			





#### Unit V

## 8. Carboxylic Acids (Aromatic and Aliphatic)

- 8.1 Introduction
- 8.2 Acidity of Carboxylic Acids
- 8.3 Effect of Substituents on Acidity
- 8.4 Inductive Effect
- 8.5 Qualitative Test for CarBoxylic Acid (R-/Ar-COOH)
- 8.6 Structures and Uses of some Carboxylic Acids and Derivatives
  - Questions

#### 9. Aliphatic Amines

- 9.1 Introduction
- 9.2 Basicity of Amines
- 9.3 Effects of Substitutents on Basicity of Amines
- 9.4 Qualitative Tests for Amines
- 9.5 Structure and Uses of some Amines
  - Questions

Index

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### Chapter ... 1

## CLASSIFICATION, NOMENCLATURE AND ISOMERISM

#### LEARNING OBJECTIVES +

- To know different types of Classes and Organic Compounds.
- To understand the IUPAC Rules for Nomenclature of Organic Compounds.
- To draw the Structure from given name.
- To give the name for given Structure.
- To understand types of Isomerism of the Organic Compound.

#### 1.1 CLASSIFICATION OF ORGANIC COMPOUNDS

Classification of organic compounds on the basis of functional group and elemental composition:

- Compounds containing carbon and hydrogen atoms only: Hydrocarbons (Alkanes, Alkenes, Alkynes, Aromatic Hydrocarbons, Arylalkyl Hydrocarbons, Alicyclic Hydrocarbons).
- Compounds containing carbon, hydrogen and oxygen atoms only: Alcohols, Phenols, Ethers, Epoxides, Carbonyl compounds, Aldehydes and Ketones, Carboxylic acids, Esters, Anhydrides.
- 3. Compounds containing Carbon, Hydrogen and Nitrogen atoms only: Amines and Imines, Nitriles, Hydrazines.
- Compounds containing Carbon, Hydrogen, Halogens with or without Oxygen: Alkyl Halides, Aryl Halides, Acyl Halides.
- Compounds containing Carbon, Hydrogen, Oxygen and Nitrogen atoms only: Amides, Imides, Aldoximes, Ketoximes, Nitro compounds.
- Compounds containing Carbon, Hydrogen and Sulphur with/without Nitrogen, Oxygen and Halogen: Sulphonic acids, Sulphonylhalides, Sulphonamides.
- IUPAC nomenclature of all classes of compounds; Nomenclature of Mono-substituted and Poly-substituted compounds. (Recent rules of IUPAC referred).

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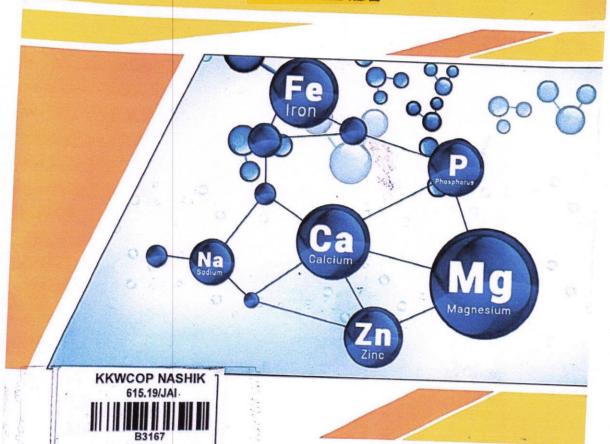
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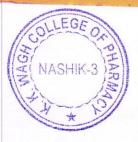
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#### Dr. K. S. Jain

M. Pharm., Ph.D., FIC
Principal & Professor, (Pharmaceutical Chemistry)
K.K. Wagh Education Society's
K.K. Wagh College of Pharmacy
Panchwati, Nashik - 422 003

#### J. N. Kadam

M. Pharm.
Asst. Professor of Pharm. Chem.
Siddhant College of Pharmacy
Sudumbare, Pune 412109

#### M. G. Shinde

M. Pharm.
Asst. Professor of Pharm. Chem.
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## Contents

1	1. Limit Tests for Various Ions - Introduction	1.1 - 1.26
	To perform Limit Test for Chloride	1.2.
	To perform modified Limit Test for Chloride.	1.4
	3. To perform Limit Test for Sulphate	1.5
*	4. To perform modified Limit Test for Sulphate	1.7
	-5. To perform Limit Test for Iron	1.8
	6. To perform Limit Test for Lead	1.11
	To perform Limit Test for Arsenic	1.15
	8. To perform modified Limit Test for Arsenic	1.17
	9. To perform Limit Test for Heavy metals	1.21
2.	Qualitative Analysis of Inorganic Substances	2.1 - 2.41
	10. Magnesium Hydroxide	2.22
	11. Ferrous Sulphate	2.25
	12. Sodium Bicarbonate	2.29
	13. Calcium Gluconate	2.33
	(14) Copper Sulphate	2.36
		n e glassina Filler.
3.	Tests for Purity	3.1 - 3.6
	15. To determine the neutralizing capacity of Aluminium Hydroxide G	iel 3.1
	16. To determine the Swelling Power of Bentonite	3.2
	17. To determine Potassium Iodate and Iodine in Potassium Iodide	3.2
4.	Preparation of Inorganic Pharmaceuticals	4.1 - 4.4
	18. To prepare Boric Acid	4.1
	19. To prepare Ferrous Sulphate	4.2
	20. To prepare Potash Alum	4.3
		4.5



### Limit Tests

Limit Tests are quantitative or semi-quantitative tests designed to identify or control small quantities of impurities. These tests should be specific and sensitive.

Limit = A value or amount that is likely to be present in a substance.

Test = To examine or to investigate

Impurity = A foreign matter present in a compound

#### **Definition:**

Limit test is defined as a quantitative or semi-quantitative test designed to identify and control small quantities of impurities which are likely to be present in the substance.

#### Importance of Limit Tests:

- 1. To find out the harmful amount of impurities
- 2. To find out avoidable / unavoidable amount of impurities.

#### Types of Limit Tests:

- 1. Comparison method
- 2. Quantitative determination
- 3. Test in which there is no visible reaction.

#### **General Principles:**

- 1. If the sample is lighter (in colour/turbidity/opalescence) than the standard solution then it is within the pharmacopoeial limit (accepted).
- 2. If the sample is darker/heavier than the standard solution then it is above the pharmacopoeial limit (rejected).
- Specificity of a Limit Test: A given limit test for a trace impurity should involve some selective reaction of the reagent with the trace impurity under consideration/detection specifically characteristic only to it.
- 4. Sensitivity of a Limit Test: As most of the limit tests involve dilute solutions and results are based on concentration of the trace impurity, the results may take longer duration to become observable or appreciable. Thus, consideration of duration of test needs to be of prime consideration in designing the limit test.

### Nessler's Cylinder (IP appendix VII A127):

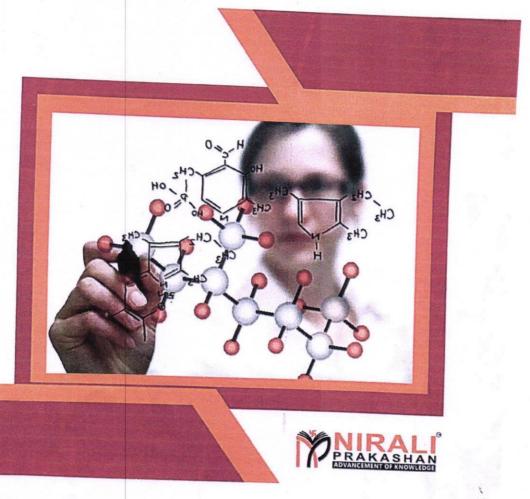
It is a clear glass cylinder with normal capacity of 50 ml. However, some Nessler's cylinders are of 100 ml capacity. The overall height is about 15 cm, the external height to the 50 ml mark is 11.0 to 12.4 cm and the thickness of the wall is around 1.0 to 1.5 mm, while, the thickness of the base is about 1.0 to 3.0 mm. The external height to the 50 mark of cylinders used for the test must not differ by more than 1 mm in the given pair.

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### Contents ...

1.	Ster	eoisomerism 1.1	- 1.34
	1.1	Introduction	1.1
	1.2	Stereoisomerism	1.3
		1.2.1 Configurational Isomerism	1.3
		1.2.2 Conformational Isomerism	1.4
	1.3	Optical Isomerism	1.4
	1.4	Optical Activity	1.5
	1.5		1.7
	1.6	Optical Isomers	1.9
		1.6.1 Enantiomers and Diastereomers	1.9
	1.7	DL System of Nomenclature of Optical Isomers	1.13
	1.8	RS System of Nomenclature of Optical Isomers	1.15
	1.9	Reactions of Chiral Molecules	1.23
		1.9.1 The Stereochemistry of SN2 and SN1 Reactions	1.23
		1.9.2 The Stereochemistry of E1 and E2 Reactions	1.24
		Racemic Modification and Resolution of Racemic Mixture	1.27
	1.11	Asymmetric Synthesis (Stereoselective Synthesis or the Synthesis of	
	100	Molecules)	1.30
		• Questions	1.32
2.	Geor	netrical Isomerism 2.1 -	2.30
	2.1	Introduction	2.1
	2.2	Nomenclature of Geometrical Isomers	2.2
	2.3	Methods of Determination of Configuration of Geometrical Isomers	2.6
		2.3.1 Chemical Methods	2.6
		2.3.2 Physical Methods	2.7
	2,4	Conformational Isomerism	2.9
		2.4.1 Newmann Projections	2.9
		2.4.2 Sawhorse Projections	2.10
		2.4.3 Conformational Isomerism and Conformational Analysis	2.10
		2.4.4 Conformations of Ethane	2.11
		2.4.5 Conformations of Butane	2.12
		2.4.6 Conformations of Cyclohexane	2.14
		2.4.7 Equatorial and Axial Bonds in Chair form of Cyclohexane	2.14
		2.4.8 Conformation of Cyclohexane - Conformational Analysis	2.20
	2.5	Stereo Isomerism in Biphenyl Compounds	
		(Atropisomerism) and Conditions for Optical Activity	2.22
		Stereoselective and Stereospecific Reactions	2.24
	2.6	Secretary to revenue and the second of the s	2.24
		Questions	2.20



	3.1 - 3.26
3. Heterocyclic Compounds - I	3.1
3.1 Introduction	3.2
3.2 Nomenclature of Heterocyclic Compounds	3.3
3.3 Classification of Heterocyclic Compounds	3.8
3.4 Pyrrole	3.8
3.4.1 Structure	3.9
3.4.2 Physical Properties 3.4.3 Synthesis of Pyrroles	3.9
3.4.4 Chemical Reactions	3.11
3.4.5 Medicinal uses of Pyrrole	3.14
3.5 Furan	3.15
3.5.1 Structure	3.15
3.5.2 Physical Properties	3.15
3.5.3 Synthesis of Furan	3.15
3.5.4 Chemical Reactions	3.16
3.5.5 Medicinal uses of Furan	3.19
3.6 Thiophene	3.20
3.6.1 Structure .	3.20
3.6.2 Physical Properties	3.21
3.6.3 Synthesis of Thiophene	3.21
3.6.4 Chemical Reactions	3.21
3.6.5 Medicinal uses of Thiophene	3.22
3.7 Relative Aromaticity and Reactivity of Pyrrole, Furan and Thiophene	3.24
Questions	3.25
4. Heterocyclic Compounds – II	4.1 - 4.46
4.1 Pyrazole	
4.1.1 Structure	4.1
4.1.2 Physical Properties	4.1
4.1.3 Synthesis of Pyrazole	4.1
4.1.4 Chemical Reactions	4.1
4.1.5 Medicinal uses of Pyrazole	4.2
4.2 IMIdazole	4.3
4.2.1 Structure	4.4
4.2.2 Physical Properties	4.4
4.2.3 Synthesis of Imidazole	4.4
4.2.4 Chemical Reactions	4.5
4.2.5 Medicinal uses of Imidazole	4.6
4.3 Oxazole	4.11
SAUZOIE SAUZOIE	4.12
4.3.1 Structure	4.12
4.3.2 Physical Properties	4.12
4.3.3 Synthesis of Oxazole	4.13
4.3.4 Medicinal uses of Oxazole	4.14
and the state of t	7.27



4.4	Thiazole	4.15
	4.4.1 Structure	4.15
	4.4.2 Physical Properties	4.16
	4.4.3 Synthesis of Thiazole	4.16
	4.4.4 Chemical Reactions	4.16
	4.4.5 Medicinal uses of Thiazoles	4.17
4.5		4.18
	4.5.1 Structure and Physical Properties	4.18
	4.5.2 Electrophilic Attack at Positions 3 and 5	4.18
	4.5.3 Synthesis of Pyridine	4.19
	4.5.4 Chemical Properties	4.20
	4.5.5 Medicinal uses of Pyridine	4.23
4.6	Quinoline	4.24
	4.6.1 Structure, Reactivity and Orientation	4.24
	4.6.2 Synthesis of Quinoline	4.25
	4.6.3 Chemical Properties	4.28
	4.6.4 Medicinal uses of Quinoline	4.31
4.7	Isoquinoline	4.31
	4.7.1 Structure	4.31
	4.7.2 Physical Properties	4.31
	4.7.3 Synthesis of Isoquinoline	4.32
	4.7.4 Chemical Reactions	4.32
	4.7.5 Medicinal uses of Isoquinoline	4.34
4.8	Acridine	4.35
	4.8.1 Structure	4.35
	4.8.2 Physical Properties	4.35
	4.8.3 Synthesis of Acridine	4.35
	4.8.4 Chemical Reactions	4.35
	4.8.5 Medicinal uses of Acridine	4.36
4.9	Indole	4.36
1	4.9.1 Structure	4.36
	4.9.2 Physical Properties	4.36
	4.9.3 Synthesis of Indole	4.37
	4.9.4 Chemical Reactions of Indole	4.39
	4.9.5 Medicinal uses of Indole	4.41
410		4.42
	Basicity of Pyridine	4.43
4.11	Pyrimidine	4.43
	4.11.1 Synthesis of Pyrimidine	4.44
	4.11.2 Pyrimidine: Biological Activity	4.44
4.12	Purine	4.44
	• Questions	4.43



			5.1 - 5.16
5.	Read	Metal Hydride reduction by Sodium Borohydride (NaBH <sub>4</sub> ) and	
	<b>5</b> 1	Manage Wildridge reduction by Social	5.1
	3.1.	Lithium Aluminium Hydride (LiAlH <sub>4</sub> )	5.4
		Clemmensen Reduction	5.6
	5.2	Wolf Kishner Reduction	5.7
	5.3		5.8
	5.4	Birch Reduction	
	5.5	Oppenauer-Oxidation	5.10
	5.6	Dakin Reaction	5.11
	5.7	Claisen-Schimdt Condensation	5.13
	5.8	Schmidt Rearrangement	5.14
	5.9	Beckmann Rearrangement	5.15
	AT 10000 1	Questions	-



### **STEREOISOMERISM**

#### **+ LEARNING OBJECTIVES +**

After completing this chapter, reader should be able to understand:

- · Optical isomerism Optical Activity
- Enantiomerism
- Diastereoisomerism
- Meso Compounds
- Elements of Symmetry
- · Chiral and Achiral Molecules
- DL System of Nomenclature of Optical Isomers, Sequence Rules
- . RS System of Nomenclature of Optical Isomers
- · Reactions of Chiral Molecules
- Racemic Modification and Resolution of Racemic Mixture
- · Asymmetric Synthesis: Partial and Absolute

#### 1.1 INTRODUCTION

Stereochemistry is a branch of organic chemistry which deals with structure of compounds in three dimensions and hence can be termed as chemistry or study of compounds with respect to the arrangements and movements of different atoms or group of atoms in space. The word is derived from Greek word (Stereos = "three"-dimensionality).

Stereochemistry also deals with stereo-isomerism and stereo-chemical reactions of organic compounds.

Founders of Stereochemistry:



Biot realized in 1815 that the solutions of many naturally occurring compounds rotate the plane of polarization of plane polarized light.



Pasteur recognized in 1850 that the optical activity was caused by an asymmetric arrangement of atoms in a molecule.



van't Hoff with Le Bel described in 1874 how the atoms of a molecule are actually arranged in space

Fig. 1.1: Founders of Stereochemistry

(1.1)





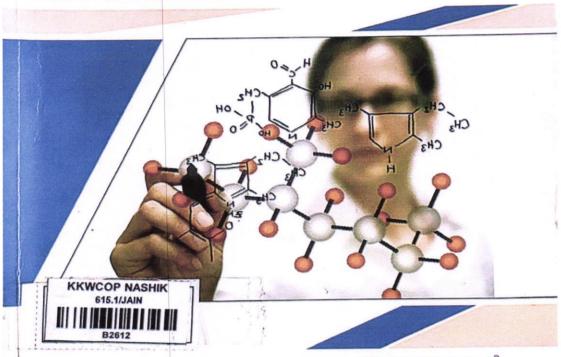
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Dr. K. S. JAIN

Dr. P. B. MINIYAR

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Contents	5 24
1. Introduction to Pharmaceutical Substances	1.1 - 1. <sub>32</sub>
2. Acids Bases and Buffers	2.1 - 2. <sub>12</sub>
3. Major Extra and Intracellular Electrolytes	3.1 - 3.20
4. Dental Products	4.1 - 4.10
5. Gastrointestinal Agents	5.1 - 5.4
6. Antacids	6.1 - 6.14
7. Cathartics	7.1 - 7.6
8. Antimicrobials	<b>8.1 - 8.1</b> 0
9. Miscellaneous Agents	9.1 - 9.14
10. Radiopharmaceuticals	10.1 - 10.14



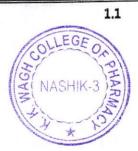
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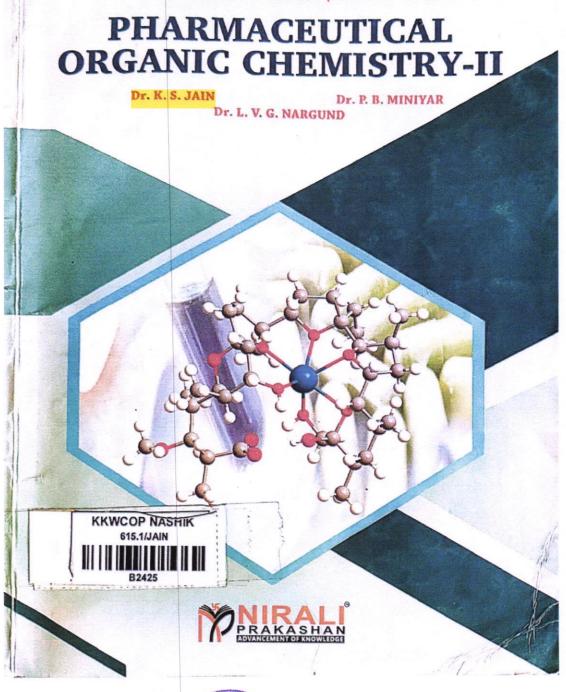
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#### Contents ...

- 1.1 Inorganic Chemistry
  - 1.1.1 Importance of Inorganic Pharmaceuticals
- 1.2 Pharmacopoeia
  - 1.2.1 History of Pharmacopoeia
  - 1.2.2 Indian Pharmacopoeia
  - 1.2.3 Indian Pharmacopoeia 2010
  - 1.2.4 Indian Pharmacopoeia 2014
  - 1.2.5 Indian Pharmacopoeia 2018
  - 1.2.6 British Pharmacopoeia
  - 1.2.7 European Pharmacopoeia
  - 1.2.8 Pharmacopoeia International (International Pharmacopoeia)
  - 1.2.9 United States Pharmacopoeia (USP)
- 1.3 Official Monograph
- 1.4 New Inclusion / Exclusion of Monograph
- 1.5 Sources of Impurities
- 1.6 Limit Tests
  - 1.6.1 Limit Test for Chlorides
  - 1.6.2 Limit Test for Sulphates
  - 1.6.3 Limit Test for Iron
  - 1.6.4 Limit Test for Lead
  - 1.6.5 Limit Test for Heavy Metals
  - 1.6.6 Limit Test for Arsenic
  - 1.6.7 Limits of Insoluble and Soluble Matter
- 1.7 Qualitative Tests for Alkali and Alkaline Earth Metals
- 1.8 Modified Limit Tests for Chlorides and Sulfates
- Question Bank



AS PER PCI REGULATIONS SECOND YEAR B. PHARM. | SEMESTER-III







# PHARMACEUTICAL ORGANIC CHEMISTRY - II

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#### Dr. K. S. Jain

M. Pharm., Ph.D., FIC
Principal & Professor
Deptt. of Pharmaceutical Chemistry
K.K. Wagh Education Society's
K.K. Wagh College of Pharmacy
Hirabai Haridas Vidyanagari, Amrut Dham, Panchavati,
Nashik 422003

#### Dr. P. B. Miniyar

M. Pharm., Ph.D., FAGE
Professor and Senate Member (SSPU)
Sinhgad Technical Education Society's
Sinhgad Institute of Pharmacy, Narhe
Pune 411 041

#### Dr. L. V. G. Nargund

M. Pharm., Ph.D., FIC
Principal and Professor
Deptt. of Pharmaceutical Chemistry
Nargund College of Pharmacy
Dattatreyanagar, Banashankari, III Stage
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### **Contents**

			Unit I		3
4	Do	nzon	and Its Derivatives	- 1.1 -	1.22
1.	1.1		cture of Benzene		1.1
	1.2		ytical Evidences	an dig of the	1.2
	1.3		hetic/Chemical Evidences	Sec. 13	1.2
	1.4	•	tal Picture of Benzene	ZH ZHÀ	1.3
	1.5		onance and Aromaticity in Benzene		1.4
	1.6		tions of Benzene		1.6
	1.7			and the second second second	1.7
	1.8		rophilic Aromatic Substitution	177 gags bets 11.8	1.17
	1.9		eophilic Aromatic Substitution cture and Uses of some selected Benzen	a Derivatives	
1	1.9			e Delivatives,	1.18
			DDT, Saccharin, BHC and Chloramine	light and the same of the same	1.22
		• (	Questions	of approximate the second	
		1.	Unit II	A to the American of State of	ja -
2.	Ph	enols		2.1'-	2.13
	2.1	Intro	duction as an amount that of	medil estila estados <b>diá</b> ns	2.1
	2.2		ity of Phenols	1 m	2.2
	2.3		nods of Preparation		2.3
	2.4		tions of Phenols	of the beautiful and the second	2.6
		• (	Questions	2.124	2.13
3.	Arc		c Amines	700 S C S C S C S C S C S C S C S C S C S	- 3.14
	3.1			er common by the officers	3.1
54-	3.2			no en eno conta A	3.2
	3.2		Reduction of Nitriles	and multiple regard queries	3.2
				mor pumpers with	3.2
1000	pre-1 5 1		Reduction of Nitrocompounds	L. L. C.	
			Selective Reduction by Ammonium Su	ipnide	3.3
		3.2.4	Reductive Amination of Ketone	and the state of	3.3
10.00		3.2.5			3.3
			Ammonolysis of Aryl Chloride		3.3
	3.3		nical Reactions of Aromatic Amines	Diphormatics of the model C	3.3
15			Formation of Amide from Acyl Chlorid	e	3.3
			Salt Formation	2000年100日 (1910年1月)	3.4
34		3.3.3	Reaction with Nitrous Acid	w net eyest and	3.4
	. 50.	3.3.4	Carbylamine Reaction	with the molificate after	3.4
			Acetylation	College Available records	
			Reaction with Aldehydes (Schiff's Base	9)	3.5
		3.3.7	Oxidation Reactions		3 5



	3.4 Basi	icity of An	nines				2.5
	3.5 Effe	cts of Sub	stitutents on Basicity o	f Amines			3.5
	3.6 Fact	ors affect	ing on Basicity	Animes	record on the	Maria Profes	3.7
Andrew Co			nce Effect				3.7
No.	3.6.2	Inductiv	e Effect		not, 41%		3.8
E.C.	3.6.3	Hydrog	en Bonding	a Aleganger +	o activitie R	110	3.8
4.7	3.6.4		Nature of Electron Pair		Petrotro P	<u>.</u> 1	3.9
9.0	3.6.5	Steric E					3.9
E5	3.6.6	Solvatio	n Effect	in a second of	and took the st		3.9
AND D	3.6.7	Electron	egativity of Bonded A	oms	1777	1.0	
	.7 Aryl l	Diazoniun	n Salts	.01113	the many		3.10
213	3.7.1	History	of Aryl Diazonium Salts		5 3.772		3.10
AL à	3.7.2	Aryl Dia	Annual Control of the		Riskinski P		3.10
			ion of Benzenediazoni	um Chlorida	V 10 2010	1.52	3.10
616	3.7.4	Syntheti	c Applications of Aryl I	Diazonium Salte	-eserciants -	A .	3.11
616	3.7.5	Importar	nce of Diazonium Salts	JIGZOTIIGITI SAILS	e e e e e e e e	197 : (W)	3.12
E1.2	• Q	uestions	and a supplied that the suppli	i annite e	aisaninys i Alemanys i	L. S	3.13
4. A		Acids		energy mount	rico - mate		3.14
IS 8 4.1		luction		SHEET HALL SHE W.	and a second	4.1 -	
15.6 4.2		y: Salt For			n entiring ner		4.1
4.3			tions of Benzoic Acid		ond draw(		4.2
4.4							4.3
		estions	paration of Aryl Carbo	xylic Acids			4.6
	· Qu	iestions		dan f	A FATER	Marin A	4.8
4		1	Unit		A CEORN		
. Fa	ts and (	Oils	The state of the s		the right of	-	56
5.1	Introdu	uction	chara.	oby? in accept	400	87	5.1
5.2	Reaction	ons of Fat	ty Acids		edic on A., Wakabari		5.2
. 5	5.2.1 H	Hydrolysis	5	n o Si dilmi			5.2
X.	5.2.2 H	Hydrogen	ation	ormation House			5.2
5.3	Rancidi	ty of Oil (	(Rancidification)		olay Jáb veli		5.2
1	5.3.4	Orying Oil	s		विद्यास <b>्ट</b>		5.3
5.3	Principl	es and Si	gnificance of various	Analytical Consta	ints aris	San party T	5.3
2	5.3.1 A	cid Value		First guestings			5.4
9	5.3.2 S	aponifica	tion Value	. William a	STATES	1.4.5 2	5.4
·	5.3.3 E	ster Value	e	rit mas yill	onsomba.	9.5.2	5.5
	5.3.4 Ic	odine Val	ue	1941 5 1981	1 1 A	e. ( ) 9	5.5
	5.3.5 A	cetyl Valu	ıe		or adopt		5.6
	5.3.6 R	eichart-M	leissl (RM) Value		renotheens		5.6
		stions	and an artist of the second of				5.6
			OLLEGE	000	1	V and its	Sales .

				0			6.	1-6.24
	6. Po	lynuclea	ar Hydrocarbons				THE REAL PROPERTY.	
	6.1	700 In 1991						6.1
	6.2	Naphth					2,5	6.1
	0.2		ynthesis of Naphthal	ene			6.9.6	6.3
1			roperties		- pa - g el		9.35	6.4
			hemical Reactions					6.6
		624 11	ses of Naphthalene a	nd its Derivative	S		1+	6.9
		6.2.5 St	ructures of Naphtha	ene Derivatives	10757	- 303		6.10
	6.3	Anthrace					1684	6.10
	£ .	6.3.1 Sy	The state of the s		rent i de		186	6.11
			emical Reactions	33.	161			6.12
			es of Anthracene and	d its Derivatives		. 17		6.14
		6.3.4 Str	uctures of Anthracel	ne Derivatives	the first of the	*31 * 1		6.15
		Phenanth	rene	grandy e is set	v	FE 197	1.1.4	6.16
		6.4.1 Syr		gradius Sq.45	apid to so a	mayeri	2.3.63	6.16
		•	emical Reactions			11.19.11	0 14	6.19
		5.4.3 Use	es of Phenanthrene a	nd its Derivative	S	985526	y Histo	
			nethane (Benzylbenz			dest et.	are of the	6.21
5.0		riphenylr			W-2 10			6.21
7.5		Questi		·	6 to 200003			6.24
4 1			A.S.A.	Unit V	orth activi		ci a M	A A
_			F1.77.274 01	· Cilic O Jr , it's				.1 - 7.7
7.	_	alkanes				, 417677 m		7.1
		troductio		and the same		100		7.1
2. 5			on of Saturated Cycl			alie	1275 1	7.3
			f Preparation of Cyc n Dihalides	ioaikaries		anima'a	u fel	7.3
			kmann Reaction		100 A 17 C 3	and the same	i kanadi	
6.0			nons-Smith Reaction			F 71	11.2	7.3
			Aromatic Hydrocar		70 12 14			
3.2			Cycloalkanes		0.000			
18			er Strain Theory					7.4
		-	-Mohr's Theory of S	Strainless Rings				
-			Cyclopropane and			850		
			itution Reactions					7.6
5 /			on of Cl <sub>2</sub> and Br <sub>2</sub>	233.				
\$ 17			on of HBr and HI			97		7.6
* '			on of Hydrogen			- post		7.6
B.C.		Question				. 30. Pr		7.7
16	•	Question	3	<b>3</b> 44	of effects	A., 6(5.3	9 62	7.0
1 37			The state of the s			the state of	1 . 6	





Chapter ... 1

## BENZENE AND ITS DERIVATIVES

#### LEARNING OBJECTIVES +

After completing this chapter, reader should be able to understand:

- Introduction to benzene.
- Analytical, Synthetic and other evidences in the derivation of structure of benzene
- Orbital picture of benzene.
- Resonance in benzene, aromatic characters, Huckel's rules.
- Reactions of benzene nitration, sulphonation, halogenation-reactivity, Friedel-Crafts alkylation - reactivity, limitations, Friedel-Crafts acylation.
- Substituents, effect of substituents on reactivity and orientation of mono substituted benzene compounds towards electrophilic substitution reaction.
- Structure and uses of DDT, Saccharin, BHC and Chloramine.

#### THE STRUCTURE OF BEILDENE

Benzene on which the study of aromatics began was discovered in 1825. However, it was not till 1866 that the Kekule's formula or structure I of benzene was known, till he proposed it. This structure of benzene is most accepted because the satisfactory answers it offers to various substitution products as compared with other four proposed structures II-V (Fig. 1.1.).

Kekule formula

Dewar formula

Fig. 1.1: Various structures proposed for benzene in earlier days

Chp 1 | 1.1



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3.3.2

2. Web links of Books/Chapter



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3.3.2 Number of books and chapters in edited volumes/books published and papers published in national/international conference proceedings per teacher during last five year

#### 2. Web-link of books.

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2	Bhamare V. G., Amrutkar R. D., Patil V, Upasani C. D.	Growing impact of herbal bioenhancers in pharmaceutical industries	International	2021-2022	9783110746792	K. K. Wagh College of Pharmacy	De-Gruyter	https://www.google.co.in /books/edition/Drug_Del ivery_Technology/CyJlE AAAQBAJ?hl=en&gbpv =1&dq=Drug+Delivery+ Technology:+Herbal+Bi oenhancers+in+Pharmac euticals+ISBN&printsec =frontcover
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